

MEC516/BME516: Fluid Mechanics I

Chapter 2: Fluid Statics Solved Buoyancy Problem

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Solved Buoyancy Problem: Weighing Low Density Materials

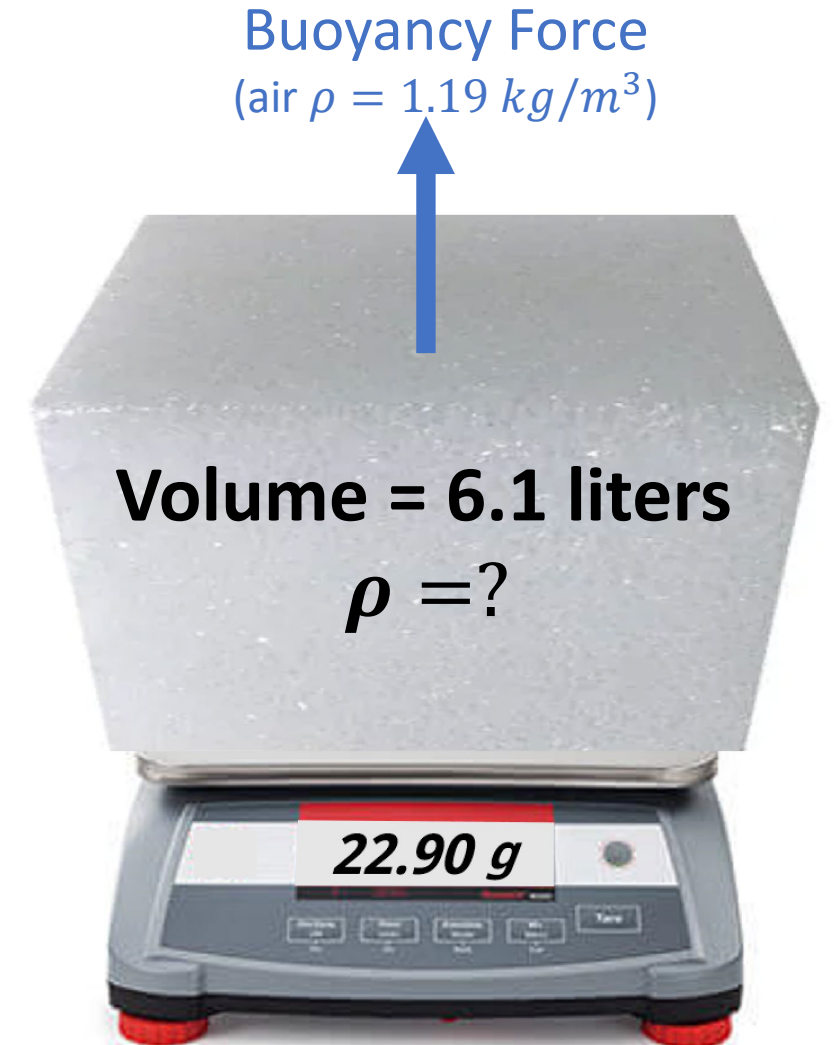
A piece of low-density foam is weighed in air at 20°C and 100 kPa. The scale indicates 22.9 grams. The volume of the foam block is 6.1 liters ($6.1 \times 10^{-3} \text{ m}^3$).

Correcting for buoyancy, what is the density (ρ) of the foam?

The density is **NOT**:

$$\rho_{foam} = \frac{0.0229 \text{ kg}}{6.1 \times 10^{-3} \text{ m}^3} = 3.75 \text{ kg/m}^3 \quad \text{X Wrong!}$$

Why not?



Buoyancy Problem: Weighing Low Density Materials

- The scale does not indicate the true mass because of buoyancy
- Force balance on the foam :

$$W = F_{scale} + F_B$$

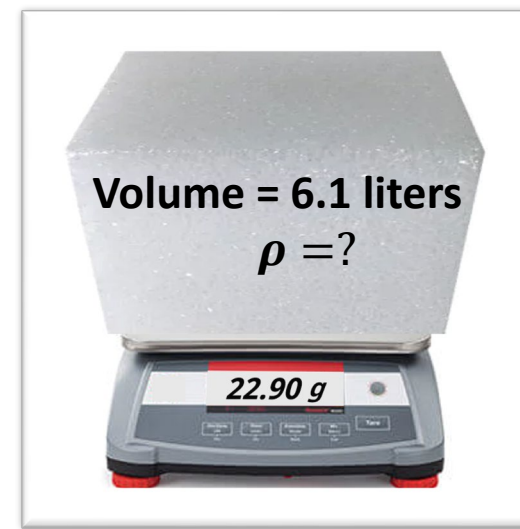
$$\rho_{foam} \forall g = m_{scale} g + \rho_{air} \forall g$$

- Solving for density of the foam:

$$\rho_{foam} = \frac{m_{scale}}{\forall} + \rho_{air} \quad (*)$$

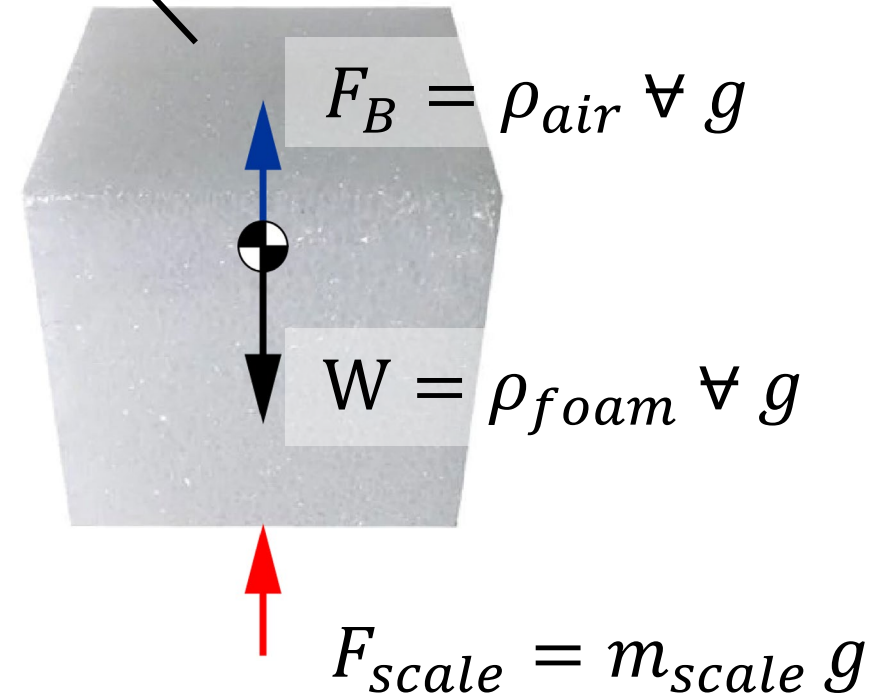
Ideal gas equation (or air properties table):

$$\rho_{air} = \frac{p}{RT} = \frac{100 \text{ kPa}}{0.287 \frac{\text{kJ}}{\text{kgK}} (20+273) \text{K}} = 1.19 \frac{\text{kg}}{\text{m}^3}$$



Foam

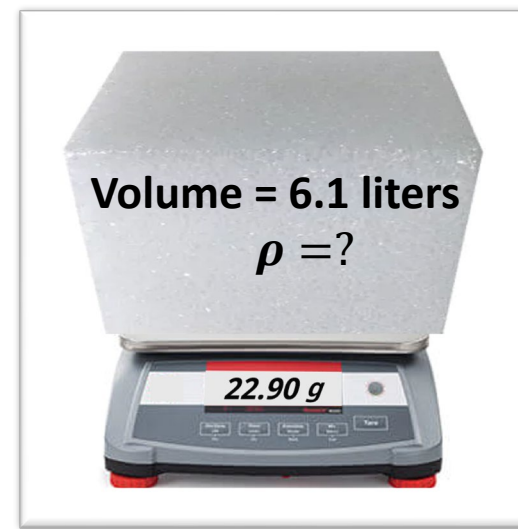
Volume \forall



Buoyancy Problem: Weighing Low Density Materials

$$\rho_{foam} = \frac{m_{scale}}{V} + \rho_{air} (*) \quad \rho_{air} = 1.19 \frac{kg}{m^3}$$

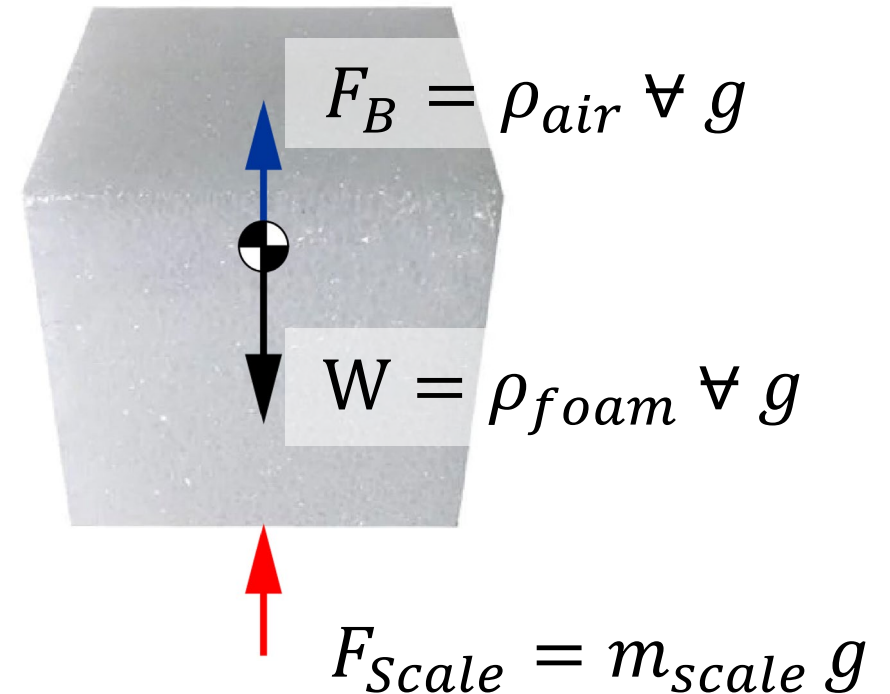
$$\rho_{foam} = \frac{0.0229 kg}{6.1 \times 10^{-3} m^3} + 1.19 \frac{kg}{m^3} = 4.94 \frac{kg}{m^3} \quad \text{Answer}$$



- If you neglect buoyancy effects:

$$\text{Percent Error} = \frac{3.75 - 4.94}{4.94} \times 100\% = -24\%$$

- A significant error for low density materials



Does buoyancy affect the reading on your bathroom scale?

Answer: Yes. $\rho_{human} \approx 1000 \text{ kg/m}^3$. Scale reads 0.12% lighter ($\sim 0.2 \text{ lb}$) than your actual weight



Actual weight requires weighting yourself in a vacuum...

END NOTES

- All the videos (and pdf downloads) for this course available at: www.drdauidnaylor.net
- Presentation prepared and delivered by Professor David Naylor

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